## organic papers

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#### Key indicators

Single-crystal X-ray study T = 293 K Mean  $\sigma$ (C–C) = 0.004 Å R factor = 0.044 wR factor = 0.127 Data-to-parameter ratio = 16.9

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

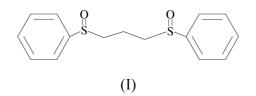
# (R,R)-1,3-Bis(phenylsulfinyl)propane

The molecular structure of the title compound,  $C_{15}H_{16}O_2S_2$ , adopts the *R*,*R* form. The dihedral angle between the two phenyl rings is 17.4 (2)° and the torsion angle between the two S=O groups is 69.3 (3)°.

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## Comment

Sulfoxides are of considerable interest because of the potential antitumor activity of their  $Pt^{II}$  complexes. Bis-sulfoxide complexes of  $Pt^{II}$  are normally *cis* and the antitumor activity depends on the chirality of the ligands (Farrell *et al.*, 1990). There are three isomers with different chirality in bis-sulfoxides, *meso*, (+)-*rac* and (-)-*rac* (Greene *et al.*, 1971). Although the stereoisomers can be characterized by NMR spectroscopy, their molecular structures are less studied (Cattalini *et al.*, 1979). In the present paper, we report the crystal structure of the title compound, namely (*R*,*R*)-1,3bis(phenylsulfinyl)propane, (I).



The molecular structure shown in Fig. 1 has the R,R conformation. The dihedral angle between the two phenyl rings is 17.4 (2)° and the torsion angle between the two S=O groups is 69.3 (3)°. A similar situation has been found in the structure of (R,R)-1,2-bis(phenylsulphinyl)propane (Cattalini *et al.*, 1979) and there is a close agreement between their geometric parameters.

### **Experimental**

Compound (I) was prepared according to a reported procedure and was characterized by NMR, IR and elemental analyses, giving results consistent with those in the literature (Zhang *et al.*, 1997). Colorless single crystals of the title compound suitable for X-ray diffraction were obtained by slow diffusion of acetone into the chloroform solution of (I).

> $D_x = 1.360 \text{ Mg m}^{-3}$ Mo  $K\alpha$  radiation Cell parameters from 6382

 $0.30 \times 0.25 \times 0.20 \text{ mm}$ 

reflections  $\theta = 2.3-26.4^{\circ}$   $\mu = 0.37 \text{ mm}^{-1}$  T = 293 (2) KPrism, colorless

Crystal data

$C_{15}H_{16}O_{2}S_{2}$
$M_r = 292.40$
Monoclinic, $P2_1/n$
a = 8.444 (4)  Å
b = 15.580 (7)  Å
c = 11.579 (6) Å
$\beta = 110.333 \ (9)^{\circ}$
$V = 1428.4 (12) \text{ Å}^3$
Z = 4

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### Data collection

Bruker SMART1000 diffractometer 2469 reflections with  $I > 2.0\sigma(I)$  $R_{\rm int} = 0.026$  $\omega$  scans  $\theta_{\rm max} = 26.4^\circ$ Absorption correction: multi-scan  $h = -10 \rightarrow 5$ [SAINT (Bruker, 1998) and SADABS (Sheldrick, 1997)]  $k = -19 \rightarrow 19$  $l = -13 \rightarrow 14$  $T_{\rm min} = 0.898, \ T_{\rm max} = 0.930$ 6457 measured reflections Intensity decay: none 2910 independent reflections

## Refinement

Refinement on  $F^2$  $R[F^2 > 2\sigma (F^2)] = 0.044$  $wR(F^2) = 0.127$ S = 1.12910 reflections 172 parameters H-atom parameters constrained

 $w = 1/[\sigma^2(F_o^2) + (0.0643P)^2]$ + 0.4887P] where  $P = (F_o^2 + 2F_c^2)/3$  $(\Delta/\sigma)_{\rm max} < 0.001$  $\Delta\rho_{\rm max} = 0.75 \text{ e } \text{\AA}^{-3}$  $\Delta \rho_{\rm min} = -0.33 \text{ e } \text{\AA}^{-3}$ Extinction correction: SHELXL Extinction coefficient: none

The C-bound H atoms were placed in geometrically calculated positions and included in the final refinement in the riding-model approximation with displacement parameters derived from the atoms to which they were bonded.

Data collection: SMART (Bruker, 1998); cell refinement: SMART; data reduction: SAINT1000 (Bruker, 1998); program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: ORTEPII (Johnson, 1976).

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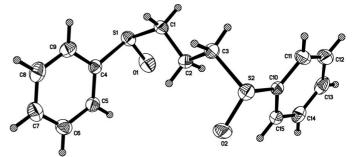


Figure 1 ORTEPII (Johnson, 1976) view of the title compound with 30% probability ellipsoids.

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## References

- Bruker (1998). SMART, SAINT and SHELXTL. Bruker AXS Inc., Madison, Wisconsin, USA.
- Cattalini, L., Michelon, G., Marangoni, G. & Pelizzi, G. (1979). J. Chem. Soc. Dalton Trans. pp. 96-101.
- Farrell, N., Kiley, D. M., Schmidt, W. & Hacker, M. P. (1990). Inorg. Chem. 29, 397-403.
- Greene, J. L., Stevlin, J. & Shevlin, P. B. (1971). J. Chem. Soc. Chem. Commun. pp. 1092-1093.
- Johnson, C. K. (1976). ORTEPII. Report ORNL-5138. Oak Ridge National Laboratory, Tennessee, USA.
- Sheldrick, G. M. (1997). SADABS, SHELXS97 and SHELXL97. University of Göttingen, Germany.
- Zhang, R.-H., Ma, B.-Q., Bu, X.-H., Wang, H.-G. & Yao, X.-K. (1997). Polyhedron, 16, 1123-1125.